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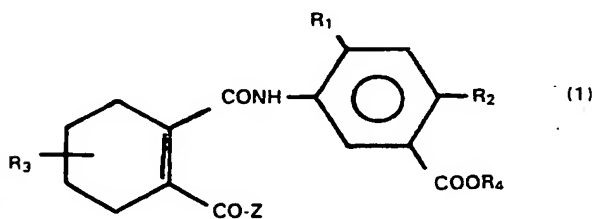
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㉜ **Herbicidal N-substituted-3,4,5,6-tetrahydrophthalamic acid derivatives.**

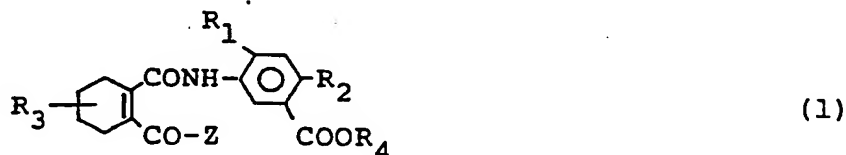
㉝ N-substituted-3,4,5,6-tetrahydrophthalamic acid derivatives of the formula:



wherein R<sub>1</sub> is hydrogen or halogen; R<sub>2</sub> is halogen; R<sub>3</sub> is hydrogen or C<sub>1</sub>-C<sub>4</sub> alkyl; R<sub>4</sub> is hydrogen or C<sub>1</sub>-C<sub>8</sub> alkyl which may be substituted by halogen or C<sub>1</sub>-C<sub>4</sub> alkoxy; and Z is hydroxy, C<sub>1</sub>-C<sub>8</sub>-primary or secondary alkylamino, C<sub>2</sub>-C<sub>5</sub>-alkenyl amino, C<sub>3</sub>-C<sub>7</sub>-cycloalkylamino, furfurylamino, benzylamino, morpholino which may be substituted by methyl, or C<sub>2</sub>-C<sub>4</sub>-alkylamino which may be substituted by phenyl, C<sub>1</sub> or C<sub>2</sub>-alkylamino, C<sub>1</sub>-C<sub>4</sub>-alkoxy, C<sub>1</sub>-C<sub>4</sub>-alkylthio or cyano; and alkali metal salts thereof: are useful as herbicides.

DESCRIPTIONTITLE: HERBICIDAL N-SUBSTITUTED-3,4,5,6-TETRAHYDRO-PHTHALAMIC ACID DERIVATIVES

The present invention relates to an N-substituted-3,4,5,6-tetrahydrophthalamic acid derivative of the formula:



wherein  $R_1$  is hydrogen or halogen,  $R_2$  is halogen,  $R_3$  is hydrogen or lower alkyl,  $R_4$  is hydrogen,  $C_1 \sim C_8$ -alkyl which may have halogen or lower alkoxy and Z is hydroxy;  $C_1 \sim C_8$ -primary or secondary alkylamino;  $C_2 \sim C_5$ -alkenyl amino;  $C_3 \sim C_7$ -alicyclic alkylamino;  $C_2 \sim C_4$ -alkylamino which may have phenyl, furfuryl,  $C_1 \sim C_2$ -alkylamino,  $C_1 \sim C_4$ -alkoxy,  $C_1 \sim C_4$ -alkylthio or cyano; morpholino; morpholino which may have methyl; or alkali metal salt of hydroxy and a herbicidal composition containing said derivative as effective component, and processes for the production thereof, a herbicidal composition

containing said compound as effective component and further to a method of killing weeds using said compound.

It is known that some N-phenyl-3,4,5,6-tetrahydrophthalamic acids have herbicidal activity. (Japanese Patent Kokai No. 44425/1973, Japanese Patent Kokai No. 96722/1973 and Japanese Patent Kokai No. 431/1974).

The present inventors found that the compound of the formula (1) shows remarkably strong herbicidal activity as compared with known compounds of the above Japanese Patent Kokais. Furthermore, the present inventors found that the compound of the formula (1) has very low phytotoxicity against crops and, therefore, they become practical herbicide.

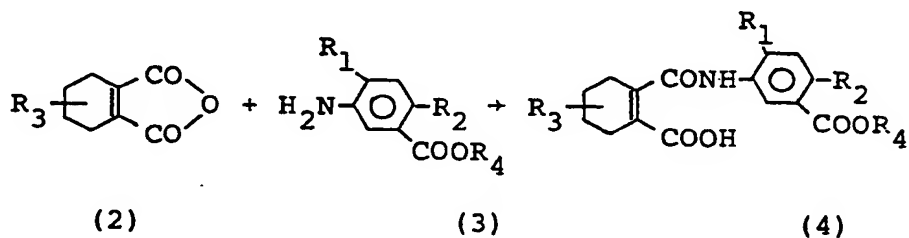
The compound of the formula (1) exhibits excellent herbicidal effect in a paddy field at a low dosage not only against annual weeds such as barnyard grasses and broadleaf weeds, but also against perennial weeds such as mizugayatsuri, bulrush, water chestnut, needle spikerush and arrowhead. The compound of the formula (1) also shows a good herbicidal effect by both pre- and post-emergence treatments in an up-land, especially against broadleaf weeds as those of amaranth, goosefoot and buckweat families at a low dosage.

On the other hand, the compound of the formula (1) is hardly phytotoxic to crops such as rice, wheat, oat, corn, soybean, cotton and sunflower.

The compound of the formula (1) of the present invention can be prepared by the processes described in a), b) and c) below.

a) The compound of the formula (1) wherein Z is hydroxy [hereinafter called "the compounds of the formula (4)"] can be obtained as follows:

3,4,5,6-tetrahydrophthalic anhydride of the formula (2) is reacted with m-aminobenzoic acid derivatives of the formula (3) in a proper solvent at a temperature of from room temperature to 120°C, preferably from room temperature to 80°C for a period of from 30 min. to several hrs. to produce the compound of the formula (4).



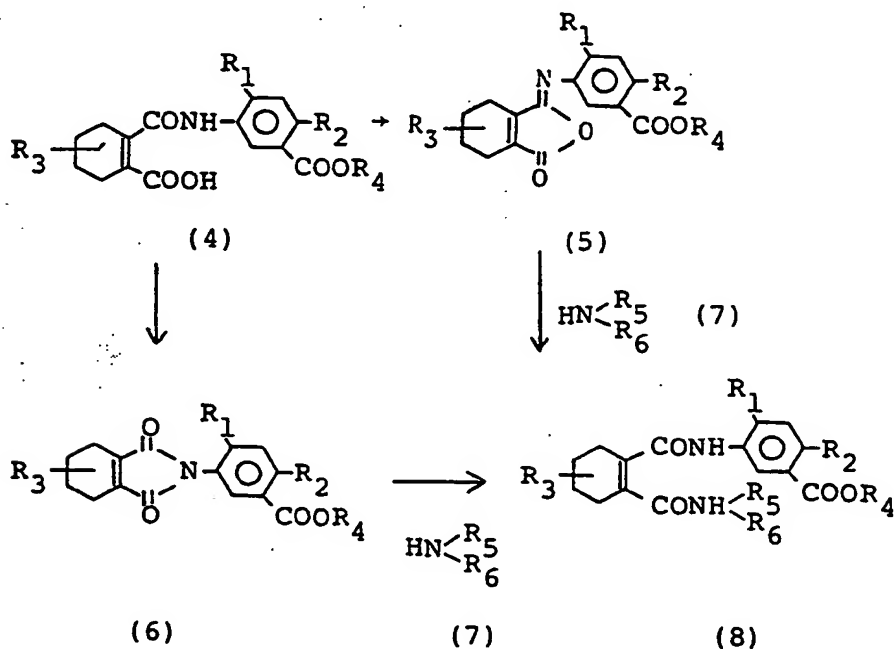
wherein R<sub>1</sub>, R<sub>2</sub>, R<sub>3</sub> and R<sub>4</sub> are defined as above.

b) The compound of formula (1) wherein Z is alkali metal salt of hydroxy is obtained by reacting the compound of the formula (4) with alkali metal hydroxide such

as sodium hydroxide and potassium hydroxide by an ordinary method.

c) The compound of the formula (1)

wherein Z is hydroxy;  $C_1 \sim C_8$ -primary or secondary alkylamino;  $C_2 \sim C_5$ -alkenylamino;  $C_3 \sim C_7$ -alicyclic alkylamino;  $C_2 \sim C_4$ -alkylamino which may have phenyl, furfuryl,  $C_1 \sim C_2$ -alkylamino  $C_1 \sim C_4$ -alkoxy,  $C_1 \sim C_4$ -alkylthio or cyano; morpholino; morpholino which may have methyl [hereinafter called "the compound of the formula (8)"] is obtained by heating the compound of the formula (4) in the presence of a proper dehydrating agent in a proper solvent at a temperature from room temperature to 100°C.



wherein  $R_1, R_2, R_3, R_4$  are defined as above,  $R_5$  and  $R_6$  are defined after.

The compound of the formula (6) is obtained by heating the compound of the formula (4) in a proper solvent, if necessary in the presence of an acid catalyst at a temperature of from 50°C to 150°C.

The compound of the formula (8) is obtained by reacting the compound of the formula (5) or (6) with the compound of the formula (7) wherein  $R_5$  is hydrogen or lower alkyl,  $R_6$  is hydrogen;  $C_1\sim C_8$ -alkyl;  $C_2\sim C_5$ -alkenyl,  $C_3\sim C_7$  alicyclic alkyl;  $C_2\sim C_4$ -alkyl which may have phenyl, furfuryl,  $C_1\sim C_2$ -alkylamino,  $C_1\sim C_4$ -alkoxy  $C_1\sim C_4$ -alkylthio or cyano; or  $R_5$  may together with  $R_6$  make morpholino which may have methyl, preferably at a temperature of from room temperature to 120°C.

As examples of the dehydrating agents, there can be mentioned carbodiimide derivatives such as dicyclohexylcarbodiimide, diethylcarboxydiimide, mixture of a base and an acid halogenating agent or acylating agent.

As examples of the bases, there can be mentioned an aliphatic, aromatic or heterocyclic tertiary amines such as triethyl amine, dimethyl aniline or pyridine, and carbonates or hydrogen carbonates of alkali metal such as sodium carbonate, potassium carbonate and sodium hydrogen carbonates.

As examples of the acid halogenating agent there can be mentioned thionyl chloride or phosphorus oxychloride.

As examples of the acylating agent there can be mentioned an acid anhydride of an organic carboxylic acid or chlorocarbonic acid ester such as acetic acid anhydride or methyl chlorocarbonate.

As examples of the proper solvents, there can be mentioned lower aliphatic acids (acetic acid, propionic acid), aromatic compounds (toluene, xylene, chlorobenzene), halogenated hydrocarbons (chloroform, carbon tetrachloride, perchloroethylene), alcohols (methanol, ethanol), ethers (diethylether, dioxane tetrahydrofuran), ketones (acetone, methylethylketone) and water.

As examples of the acid catalysts, there can be mentioned p-toluenesulfonic acid, sulfuric acid and hydrogen chloride.

As examples of the halogen in  $R_1$ ,  $R_2$  and  $R_4$  there can be mentioned chloro, bromo or fluoro.

As examples of the lower alkyls in  $R_3$ , there can be mentioned methyl, ethyl, propyl or butyl.

As examples of  $C_1$  to  $C_8$  - alkyls in  $R_4$ , there can be mentioned methyl, ethyl, n-propyl, i-propyl, n-butyl, sec-butyl, tert-butyl, n-pentyl, neopentyl, n-hexyl, i-hexyl, n-heptyl, n-octyl or 2-ethylhexyl.


As examples of  $C_1 \sim C_8$ -alkyls which are substituted by halogen, there can be mentioned chloromethyl, chloroethyl or 1-fluoromethyl-2-fluoropropyl.

As examples of  $C_1 \sim C_8$ -alkyls which may have lower alkoxy, there can be mentioned methoxyethyl, ethoxyethyl or methoxybutyl.

As the examples of  $C_1 \sim C_8$ -primary or secondary alkylamino there can be mentioned methylamino, ethylamino, n-propylamino, i-propylamino, n-butylamino, i-butylamino, sec-butylamino, tert-butylamino, i-amylamino, n-hexylamino, n-heptylamino, n-octylamino, dimethylamino, diethylamino, dipropylamino, isobutylamino, methylethylamino, ethylbutylamino or propylbutylamino.

As examples of  $C_2 \sim C_5$ -alkenylamino, there can be mentioned allylamino or alkylmethylamino.

As examples of  $C_3 \sim C_7$ -alicyclic alkylamino, there can be mentioned cyclohexylamino.

As examples of  $C_2 \sim C_4$ -alkylamino which may have phenyl there can be mentioned 1,1-dimethyl-1-phenylamino or benzylamino. As examples of  $C_2 \sim C_4$  alkylamino which has furfuryl, there can be mentioned  $-NHCH_2-$   .

As example of  $C_2 \sim C_4$ -alkylamino which has  $C_1 \sim C_2$ -alkylamino, there can be mentioned  $-NH(CH_2)_3N(CH_3)_2$  .



As examples of  $C_2 \sim C_4$ -alkylaminos which have  $C_1 \sim C_4$ -alkoxy, there can be mentioned 2-ethoxyethylamino, 3-methoxypropylamino, 4-methoxybutylamino, 5-methoxybutylamino, 2-ethoxyethylamino, 3-ethoxypropylamino, 2-propoxyethylamino, 3-propoxypropylamino, 2-butoxyethylamino or 1-ethyl-2-methoxyethylamino.

As examples of  $C_2 \sim C_4$ -alkylaminos which have  $C_1 \sim C_4$ -alkylthio, there can be mentioned  $-NHCH_2CH_2CH_2SCH_3$  or  $-NHCH_2CH_2CH_2SC_2H_5$ .

As example of  $C_2 \sim C_4$ -alkylamino which have cyano, there can be mentioned cyanoethyl.

As examples of the alkali metal salts of hydroxy, there can be mentioned  $-ONa$  and  $-OK$ .

As preferable compounds in the present invention, there can be mentioned those of the formula (1) wherein  $R_1$  is chloro or fluoro,  $R_2$  is chloro or bromo,  $R_3$  is hydrogen or methyl,  $R_4$  is  $C_2 \sim C_3$ -alkyl,  $Z$  is hydroxy,  $C_1 \sim C_3$ -alkylamino or  $C_1 \sim C_3$ -alkoxy- $C_2 \sim C_3$ -alkylamino.

As more preferable compounds in the present invention, there can be mentioned those of the formula (1) wherein  $R_1$  is fluoro,  $R_2$  is chloro or bromo,  $R_3$  is hydrogen,  $R_4$  is i-propyl,  $Z$  is  $C_1 \sim C_3$  alkylamino or  $C_1 \sim C_3$ -alkoxy- $C_2 \sim C_3$  alkylamino. Examples of such compounds include those which are given in the table below

as compounds Nos. 43, 45, 57, 62~65, 73, 74, 76~81, 83, 84, 88, 89, 91~94, 96, 97, 111.

The present invention will be illustrated in the following examples.

Synthesis Example 1: N-(4-chloro-3-isopropoxycarbonylphenyl)-3,4,5,6-tetrahydrophthalamic acid (No. 13).

24.5 g (0.16 mol) of 3,4,5,6-tetrahydrophthalic anhydride were dissolved in 200 ml of ethyl ether.

Then 32.8 g (0.15 mol) of 4-chloro-3-isopropoxycarbonylaniline were added slowly to the solution under stirring and the reaction was effected at 20 to 28°C for 2 hrs.

The resulting crystals were filtered and washed with ether and acetone and 50 g of white crystals were obtained.

M.p. 133 to 134°C

Elementary analysis:  $C_{18}H_{20}ClNO_5$

Calculated : C:59.10 H:5.51 N:3.83

Found : C:59.15 H:5.48 N:3.86

Synthesis Example 2: N-(4-bromo-2-fluoro-5-isopropoxycarbonylphenyl)-3,4,5,6-tetrahydrophthalamic acid (No. 61)

80 ml of ether were added to 15.2 g (0.1 mol) of 3,4,5,6-tetrahydrophthalic anhydride and 27.6 g

(0.1 mol) of 4-bromo-2-fluoro-5-isopropoxycarboxyaniline were added to the mixture slowly at room temperature.

After the reaction was effected for 2.5 hrs at 30 to 40°C, 50 ml of an aqueous solution containing 11 g (0.1 mol) of sodium carbonate was added to the reaction mixture at 15 to 20°C and the reaction mixture was stirred for 2 hrs.

The reaction mixture was subjected to extraction using toluene, and the toluene layer was washed with an aqueous solution of sodium carbonate. The water layers were collected and acidified by hydrochloric acid aqueous solution to produce 35 g of pale yellow crystals.

M.p. 94 to 96°C

Elementary analysis:  $C_{18}H_{19}BrFNO_5$

Calculated : C:50.48 H:4.47 N:3.27

Found : C:50.55 H:4.49 N:3.30

Synthesis Example 3: N-ethyl-N'-(4-chloro-3-isopropoxycarbonylphenyl)-3,4,5,6-tetrahydrophthalamide (No. 14)

27.22 g of N-(4-chloro-3-isopropoxycarbonylphenyl)-3,4,5,6-tetrahydrophthalamic acid (No. 13) which were obtained according to Synthesis Example 1 were suspended in 200 ml of benzene and 18.42 g (0.089 mol) of

dicyclohexylcarbodiimide were added to the suspension, and the whole was stirred at room temperature for 2 hrs. The residue which did not dissolve in benzene were removed by filtration, the filtrate was concentrated under reduced pressure to produce an oil. A small amount of n-hexane were added to the oil and the whole was allowed to stand to precipitate the crystals. 21.2 g of N-(4-chloro-3-isopropoxycarbonylphenyl)-3,4,5,6-tetrahydroisophthalimide were obtained after recrystallization of the precipitated crystals. M.p. 100 to 102°C.

To 3 g (0.009 mol) of thus obtained compound were added 20 ml of acetone, 1 g (0.015 mol) of 70% ethylamine aqueous solution was added dropwise thereto and the reaction was effected at room temperature for 2 hrs. to produce crystals. The crystals were filtered and washed with n-hexane to obtain 2.8 g of intended white crystals. M.p. 161 to 163°C.

Elementary analysis:  $C_{20}H_{25}ClN_2O_4$

Calculated : C:61.14 H:6.41 N:7.13

Found : C:61.22 H:6.43 N:7.18

Synthesis Example 4: N-methyl-N'-(4-bromo-2-fluoro-5-isopropoxycarbonylphenyl)-3,4,5,6-tetrahydrophthalamide (No. 62)

30 g (0.07 mol) of N-(4-bromo-2-fluoro-5-isopropoxy-carboxyphenyl)-3,4,5,6-tetrahydrophthalamic acid (No. 64) which had been obtained by Synthesis Example 2 were suspended in 120 ml of benzene, and 15.9 g of dicyclohexylcarboxydiimide were added to the suspension and the suspension was stirred for 2 hrs. at room temperature.

Then the residue which was not dissolved in benzene was removed by filtration and the filtrate was concentrated under a reduced pressure to obtain 30 g of crystalline N-(4-bromo-2-fluoro-5-isopropoxycarbonylphenyl)-3,4,5,6-tetrahydrophthalimide.

The pale yellow crystals which were obtained by recrystallization had m.p. of 81 to 82°C.

To 1.6 g (0.004 mol) of thus obtained compound, added 30 ml of ether and 0.3 g (0.004 mol) of methylamine using 40% of an aqueous methylamine solution was added dropwise to the mixture at room temperature under stirring for 30 minutes.

Then the precipitated crystals were obtained by filtration and washed with ether and 13.7 g of the intended compound. M.p. 122.5 to 123.5°C.

Elementary analysis:  $C_{19}H_{22}BrFN_2O_4$

Calculated : C:51.71 H:5.02 N:6.35

Found : C:51.80 H:5.08 N:6.40

Synthesis Example 5: N-methyl-N'-(4-chloro-2-fluoro-5-isopropoxycarbonylphenyl)-3,4,5,6-tetrahydrophthalamide (No. 57)

4 g of N-(4-chloro-2-fluoro-5-isopropoxycarbonylphenyl)-3,4,5,6-tetrahydroisophthalimide were dissolved in 70 ml of tetrachloromethane and to the mixture, 0.93 g of 40% of an aqueous methylamine solution was added dropwise. After the stirring for 30 minutes, the crystals were precipitated and the crystals were filtered and washed with n-hexane to produce 3.8 g of the intended compound (white crystals). M.p.: 147 to 149°C.

Elementary analysis:  $C_{19}H_{22}ClFN_2O_4$

Calculated : C:57.50 H:5.59 N:7.06

Found : C:57.72 H:5.61 N:7.10

Synthesis Example 7: N-methyl-N'-(4-chloro-2-fluoro-5-isopropoxycarbonylphenyl)-3,4,5,6-tetrahydrophthalimide (No. 57)

To 33.5 g of 3,4,5,6-tetrahydrophthalic anhydride, 200 ml of toluene, 51 g of 5-amino-2-chloro-4-fluorobenzoic acid isopropylester and 1.8 g of p-toluenesulfonic acid were added. The mixture was refluxed to effect dehydration reaction.

After the reaction, ethylacetate was added to the reaction mixture. Then the reaction mixture was washed with water and concentrated to precipitate the

crystals. The crystals were subjected to recrystallization to obtain 73.5 g of white crystalline N-(4-chloro-2-fluoro-5-isopropoxycarbonylphenyl)-3,4,5,6-tetrahydrophthalimide (m.p. 74-5°C).

4 g of thus obtained compound were dissolved in 30 ml of benzene, and then 0.97 g of 40% of an aqueous methylamine solution was added dropwise to the mixture and the mixture was stirred for 30 minutes. The precipitated crystals were filtered and washed with benzene to produce 3.8 g of white crystals having m.p. of 147 to 149°C. The data concerning IR and NMR of the compound were identical with the data of the compound obtained by Synthesis Example 5.

Examples of those compounds obtained by the above mentioned method are given in Table 1 below.

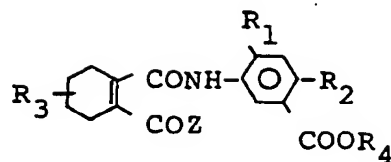

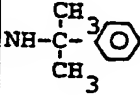
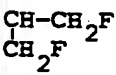
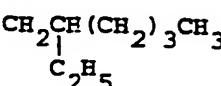



Table 1


No.	R <sub>1</sub>	R <sub>2</sub>	R <sub>3</sub>	R <sub>4</sub>	Z	M.P. (°C) or Refractive index	Appearance
1	H	F	H	H	OH	226-230	White crystal
2	H	F	H	C <sub>2</sub> H <sub>5</sub>	OH	115-116	White crystal
3	H	Cl	H	CH <sub>3</sub>	OH	123-125	White crystal
4	H	Cl	H	C <sub>2</sub> H <sub>5</sub>	NHCH <sub>3</sub>	103-106	White crystal
5	H	Cl	H	C <sub>2</sub> H <sub>5</sub>	NHC <sub>2</sub> H <sub>5</sub>	130-133	White crystal
6	H	Cl	H	C <sub>2</sub> H <sub>5</sub>	NHC <sub>3</sub> H <sub>7</sub> (i)	114-115	White crystal
7	H	Cl	H	C <sub>2</sub> H <sub>5</sub>	NHC <sub>3</sub> H <sub>7</sub> (i)	167-170	White crystal
8	H	Cl	H	C <sub>2</sub> H <sub>5</sub>	NHCH <sub>2</sub> CH=CH <sub>2</sub>	120-123	White crystal
9	H	Cl	H	C <sub>2</sub> H <sub>5</sub>	NHC <sub>4</sub> H <sub>9</sub> (n)	112-114	White crystal
10	H	Cl	H	C <sub>2</sub> H <sub>5</sub>	NHC <sub>4</sub> H <sub>9</sub> (i)	114-117	White crystal
11	H	Cl	H	C <sub>2</sub> H <sub>5</sub>	NHC <sub>4</sub> H <sub>9</sub> (s)	171-174	White crystal
12	H	Cl	H	C <sub>2</sub> H <sub>5</sub>	NH-	164-167	White crystal
13	H	Cl	H	C <sub>3</sub> H <sub>7</sub> (i)	OH	133-134	White crystal
14	H	Cl	H	C <sub>3</sub> H <sub>7</sub> (i)	NHC <sub>2</sub> H <sub>5</sub>	161-163	White crystal



No.	R <sub>1</sub>	R <sub>2</sub>	R <sub>3</sub>	R <sub>4</sub>	Z	M.P. (°C) or Refractive index	Appearance
15	H	Cl	H	C <sub>3</sub> H <sub>7</sub> (i)	NHC <sub>3</sub> H <sub>7</sub> (n)	126-129	White crystal
16	H	Cl	H	C <sub>3</sub> H <sub>7</sub> (i)	NHC <sub>3</sub> H <sub>7</sub> (i)	161-163	White crystal
17	H	Cl	H	C <sub>3</sub> H <sub>7</sub> (i)	NHC <sub>4</sub> H <sub>9</sub> (t)	205-208	White crystal
18	H	Cl	H	C <sub>3</sub> H <sub>7</sub> (i)	NHC <sub>8</sub> H <sub>17</sub> (n)	115-117	White crystal
19	H	Cl	H	C <sub>3</sub> H <sub>7</sub> (i)		145-146	White crystal
20	H	Cl	H	C <sub>3</sub> H <sub>7</sub> (i)		225-228	White crystal
21	H	Cl	H	C <sub>4</sub> H <sub>9</sub> (s)	NHC <sub>2</sub> H <sub>5</sub>	136-139	White crystal
22	H	Cl	H	C <sub>4</sub> H <sub>9</sub> (s)	NHC <sub>3</sub> H <sub>7</sub> (i)	148-151	White crystal
23	H	Cl	H	C <sub>4</sub> H <sub>9</sub> (s)	NHC <sub>3</sub> H <sub>7</sub> (n)	143-145	White crystal
24	H	Cl	H	C <sub>4</sub> H <sub>9</sub> (s)	NHC <sub>6</sub> H <sub>13</sub> (n)	103-106	White crystal
25	H	Cl	H	C <sub>2</sub> H <sub>4</sub> OCH <sub>3</sub>	OH	97-98	White crystal
26	H	Cl	H	C <sub>2</sub> H <sub>4</sub> OCH <sub>3</sub>	NHCH <sub>3</sub>	114-117	White crystal
27	H	Cl	H	C <sub>2</sub> H <sub>4</sub> OCH <sub>3</sub>	NHC <sub>2</sub> H <sub>5</sub>	140-143	White crystal
28	H	Cl	H	C <sub>2</sub> H <sub>4</sub> OCH <sub>3</sub>	NHC <sub>3</sub> H <sub>7</sub> (n)	128-132	White crystal
29	H	Cl	H		NHCH <sub>3</sub>	128-131	White crystal
30	H	Cl	H		NHC <sub>3</sub> H <sub>7</sub> (n)	$n_D^{25}$ 1.5365	Yellow oil
31	H	Cl	4- or 5-CH <sub>3</sub> mix.	C <sub>3</sub> H <sub>7</sub> (i)	OH	123-124	White crystal

No.	R <sub>1</sub>	R <sub>2</sub>	R <sub>3</sub>	R <sub>4</sub>	Z	M.P. (°C) or Refractive index	Appearance
32	H	Cl	4- or 5-CH <sub>3</sub> mix.	C <sub>3</sub> H <sub>7</sub> (i)	NHC <sub>2</sub> H <sub>5</sub>	130-132	White crystal
33	H	Cl	3- or 6-CH <sub>3</sub> mix.	C <sub>3</sub> H <sub>7</sub> (i)	NHC <sub>2</sub> H <sub>5</sub>	165-166.5	White crystal
34	Cl	Cl	H	CH <sub>3</sub>	OH	124-126	White crystal
35	Cl	Cl	H	CH <sub>3</sub>	NHCH <sub>3</sub>	170-174 dec	White crystal
36	Cl	Cl	H	CH <sub>3</sub>	NHC <sub>2</sub> H <sub>5</sub>	171-172 dec	White crystal
37	Cl	Cl	H	CH <sub>3</sub>	NHCH <sub>2</sub> CH=CH <sub>2</sub>	169 dec	White crystal
38	Cl	Cl	H	CH <sub>3</sub>	NHC <sub>4</sub> H <sub>9</sub> (s)	182-183	White crystal
39	Cl	Cl	H	CH <sub>3</sub>	NH-(CH <sub>2</sub> ) <sub>3</sub> N <sup>CH<sub>3</sub></sup> <sub>CH<sub>3</sub></sub>	99-102	White crystal
40	Cl	Cl	H	CH <sub>3</sub>	NHC <sub>3</sub> H <sub>7</sub> (i)	181-182	White crystal
41	Cl	Cl	H	CH <sub>3</sub>	NHC <sub>4</sub> H <sub>9</sub> (i)	168.5-170	White crystal
42	Cl	Cl	H	C <sub>3</sub> H <sub>7</sub> (i)	OH	109-110	White crystal
43	Cl	Cl	H	C <sub>3</sub> H <sub>7</sub> (i)	NHC <sub>2</sub> H <sub>5</sub>	139-141	White crystal
44	Cl	Cl	H	C <sub>3</sub> H <sub>7</sub> (i)	NHC <sub>3</sub> H <sub>7</sub> (n)	123-128	White crystal
45	Cl	Cl	H	C <sub>3</sub> H <sub>7</sub> (i)	NHC <sub>3</sub> H <sub>7</sub> (i)	142-145	White crystal
46	Cl	Cl	H	C <sub>3</sub> H <sub>7</sub> (i)	NHCH <sub>2</sub> CH=CH <sub>2</sub>	134.5-135	White crystal
47	Cl	Cl	H	C <sub>3</sub> H <sub>7</sub> (i)	NHC <sub>4</sub> H <sub>9</sub> (s)	157-160	White crystal
48	Cl	Cl	H	C <sub>3</sub> H <sub>7</sub> (i)	NH- 	183-185	White crystal
49	H	Br	H	C <sub>2</sub> H <sub>5</sub>	OH	64-66	Pale brown crystal

No.	R <sub>1</sub>	R <sub>2</sub>	R <sub>3</sub>	R <sub>4</sub>	Z	M.P. (°C) or Refractive index	Appearance
50	H	Br	H	C <sub>3</sub> H <sub>7</sub> (i)	N $\begin{smallmatrix} \text{CH}_3 \\ \text{CH}_3 \end{smallmatrix}$	120-122	White crystal
51	H	Br	4- or 5-CH <sub>3</sub>	C <sub>3</sub> H <sub>7</sub> (i)	OH	123-125	Pale yellow crystal
52	H	Br	4- or 5-CH <sub>3</sub>	C <sub>3</sub> H <sub>7</sub> (i)	NHC <sub>2</sub> H <sub>5</sub>	121-123	White crystal
53	Cl	Cl	4- or 5-CH <sub>3</sub>	C <sub>3</sub> H <sub>7</sub> (i)	OH	114-115	White crystal
54	Cl	Cl	4- or 5-CH <sub>3</sub>	C <sub>3</sub> H <sub>7</sub> (i)	NHC <sub>2</sub> H <sub>5</sub>	164-167	White crystal
55	Cl	Cl	4- or 5-CH <sub>3</sub>	C <sub>3</sub> H <sub>7</sub> (i)	NHC <sub>3</sub> H <sub>7</sub> (i)	158-162	White crystal
56	F	Cl	H	C <sub>3</sub> H <sub>7</sub> (i)	OH	91-93	White crystal
57	F	Cl	H	C <sub>3</sub> H <sub>7</sub> (i)	NHCH <sub>3</sub>	147-149	White crystal
58	F	Cl	H	C <sub>3</sub> H <sub>7</sub> (i)	NHC <sub>3</sub> H <sub>7</sub> (n)	136-137	White crystal
59	F	Cl	H	C <sub>3</sub> H <sub>7</sub> (i)	$\begin{smallmatrix} \text{CH}_3 \\ \text{N} \diagup \text{O} \diagdown \\ \text{CH}_3 \end{smallmatrix}$	110-113	White crystal
60	F	Br	H	H	OH	239 dec	White crystal
61	F	Br	H	C <sub>3</sub> H <sub>7</sub> (i)	OH	94-96	Pale yellow crystal
62	F	Br	H	C <sub>3</sub> H <sub>7</sub> (i)	NHCH <sub>3</sub>	122.5-123.5	White crystal
63	F	Br	H	C <sub>3</sub> H <sub>7</sub> (i)	NHC <sub>2</sub> H <sub>5</sub>	120-121	White crystal
64	F	Br	H	C <sub>3</sub> H <sub>7</sub> (i)	NHC <sub>3</sub> H <sub>7</sub> (n)	122-123	White crystal
65	F	Br	H	C <sub>3</sub> H <sub>7</sub> (i)	NHC <sub>3</sub> H <sub>7</sub> (i)	128-129	White crystal
66	F	Br	H	C <sub>3</sub> H <sub>7</sub> (i)	NHC <sub>4</sub> H <sub>9</sub> (n)	115-115.5	White crystal
67	F	Br	H	C <sub>3</sub> H <sub>7</sub> (i)	NHC <sub>4</sub> H <sub>9</sub> (s)	147.5-148	White crystal

No.	R <sub>1</sub>	R <sub>2</sub>	R <sub>3</sub>	R <sub>4</sub>	Z	M.P. (°C) or Refractive index	Appearance
68	F	Br	H	C <sub>3</sub> H <sub>7</sub> (i)	NHC <sub>4</sub> H <sub>9</sub> (t)	69.5-70.5	White crystal
69	F	Br	H	C <sub>3</sub> H <sub>7</sub> (i)	NHCH <sub>2</sub> 	106-108	White crystal
70	F	Br	H	C <sub>3</sub> H <sub>7</sub> (i)	NHCH <sub>2</sub> CH=CH <sub>2</sub>	113.5-115	White crystal
71	F	Cl	4- or 5-CH <sub>3</sub>	C <sub>3</sub> H <sub>7</sub> (i)	OH	110-113	White crystal
72	F	Cl	4- or 5-CH <sub>3</sub>	C <sub>3</sub> H <sub>7</sub> (i)	NHCH <sub>3</sub>	112-115	White crystal
73	F	Cl	4- or 5-CH <sub>3</sub>	C <sub>3</sub> H <sub>7</sub> (i)	NHC <sub>2</sub> H <sub>5</sub>	170-172	White crystal
74	F	Cl	4- or 5-CH <sub>3</sub>	C <sub>3</sub> H <sub>7</sub> (i)	NHC <sub>3</sub> H <sub>7</sub> (n)	146-149	White crystal
75	F	Br	H	C <sub>3</sub> H <sub>7</sub> (i)	ONa	> 260	Pale yellow crystal
76	F	Cl	H	C <sub>3</sub> H <sub>7</sub> (i)	NH-C <sub>2</sub> H <sub>4</sub> OCH <sub>3</sub>	87-92	White crystal
77	F	Cl	H	C <sub>3</sub> H <sub>7</sub> (i)	NHC <sub>2</sub> H <sub>4</sub> OC <sub>2</sub> H <sub>5</sub>	127-129	White crystal
78	F	Br	H	C <sub>3</sub> H <sub>7</sub> (i)	NHC <sub>2</sub> H <sub>4</sub> OCH <sub>3</sub>	114-115.5	White crystal
79	F	Br	H	C <sub>3</sub> H <sub>7</sub> (i)	NH-C <sub>2</sub> H <sub>4</sub> OC <sub>2</sub> H <sub>5</sub>	104-107	White crystal
80	F	Cl	H	C <sub>3</sub> H <sub>7</sub> (i)	NH-(CH <sub>2</sub> ) <sub>3</sub> OC <sub>2</sub> H <sub>5</sub>	75-80	White crystal
81	F	Cl	H	C <sub>3</sub> H <sub>7</sub> (i)	NH(CH <sub>2</sub> ) <sub>3</sub> OC <sub>3</sub> H <sub>7</sub> (i)	78-83	White crystal
82	F	Cl	H	C <sub>3</sub> H <sub>7</sub> (i)	NH-CHCH <sub>2</sub> OCH <sub>3</sub> C <sub>2</sub> H <sub>5</sub>	157-158	White crystal
83	F	Br	H	C <sub>2</sub> H <sub>5</sub>	NH(CH <sub>2</sub> ) <sub>2</sub> OCH <sub>3</sub>	127-128	White crystal
84	F	Br	H	C <sub>2</sub> H <sub>5</sub>	NH(CH <sub>2</sub> ) <sub>2</sub> OC <sub>2</sub> H <sub>5</sub>	120-121	White crystal
85	H	Cl	H	C <sub>3</sub> H <sub>7</sub> (i)	NH(CH <sub>2</sub> ) <sub>2</sub> OCH <sub>3</sub>	137-138	White crystal

No.	R <sub>1</sub>	R <sub>2</sub>	R <sub>3</sub>	R <sub>4</sub>	Z	M.P. (°C) or Refractive index	Appearance
86	H	Cl	H	C <sub>3</sub> H <sub>7</sub> (i)	NH(CH <sub>2</sub> ) <sub>2</sub> OC <sub>2</sub> H <sub>5</sub>	117-119	White crystal
87	H	Cl	H	C <sub>3</sub> H <sub>7</sub> (i)	NHCHCH <sub>2</sub> OCH <sub>3</sub> C <sub>2</sub> H <sub>5</sub>	121-125	White crystal
88	F	Cl	4-CH <sub>3</sub> or 5-CH <sub>3</sub> mix.	C <sub>3</sub> H <sub>7</sub> (i)	NH(CH <sub>2</sub> ) <sub>2</sub> OCH <sub>3</sub>	102-105	White crystal
89	F	Cl	4-CH <sub>3</sub> or 5-CH <sub>3</sub> mix.	C <sub>3</sub> H <sub>7</sub> (i)	NH(CH <sub>2</sub> ) <sub>2</sub> OC <sub>2</sub> H <sub>5</sub>	106-110	White crystal
90	F	Br	4-CH <sub>3</sub> or 5-CH <sub>3</sub> mix.	C <sub>3</sub> H <sub>7</sub> (i)	NHCHCH <sub>2</sub> OCH <sub>3</sub> C <sub>2</sub> H <sub>5</sub>	141-145	White crystal
91	F	Br	4-CH <sub>3</sub> or 5-CH <sub>3</sub> mix.	C <sub>3</sub> H <sub>7</sub> (i)	NH(CH <sub>2</sub> ) <sub>2</sub> OCH <sub>3</sub>	103-108	White crystal
92	F	Br	4-CH <sub>3</sub> or 5-CH <sub>3</sub> mix.	C <sub>3</sub> H <sub>7</sub> (i)	NH(CH <sub>2</sub> ) <sub>2</sub> OC <sub>2</sub> H <sub>5</sub>	99-103	White crystal
93	Cl	Cl	H	C <sub>3</sub> H <sub>7</sub> (i)	NH(CH <sub>2</sub> ) <sub>2</sub> OCH <sub>3</sub>	114-118	White crystal
94	F	Cl	H	CHCH <sub>2</sub> -CHCH <sub>3</sub> CH <sub>3</sub> CH <sub>3</sub>	NH(CH <sub>2</sub> ) <sub>2</sub> OCH <sub>3</sub>	102-106	White crystal
95	F	Cl	H	C <sub>3</sub> H <sub>7</sub> (i)	NH(CH <sub>2</sub> ) <sub>2</sub> OC <sub>3</sub> H <sub>7</sub> (i)	142-143	White crystal
96	F	Cl	H	C <sub>3</sub> H <sub>7</sub> (i)	NH(CH <sub>2</sub> ) <sub>2</sub> OC <sub>3</sub> H <sub>7</sub> (n)	129-131	White crystal
97	F	Cl	4-CH <sub>3</sub> or 5-CH <sub>3</sub> mix.	C <sub>3</sub> H <sub>7</sub> (i)	NH(CH <sub>2</sub> ) <sub>2</sub> OC <sub>3</sub> H <sub>7</sub> (n)	145-146	White crystal
98	F	Cl	4-CH <sub>3</sub> or 5-CH <sub>3</sub> mix.	C <sub>3</sub> H <sub>7</sub> (i)	NH(CH <sub>2</sub> ) <sub>2</sub> OC <sub>3</sub> H <sub>7</sub> (i)	130-132	White crystal
99	F	Cl	H	C <sub>3</sub> H <sub>7</sub> (i)	NHC <sub>4</sub> H <sub>9</sub> (sec)	163-164	White crystal

No.	R <sub>1</sub>	R <sub>2</sub>	R <sub>3</sub>	R <sub>4</sub>	Z	M.P. (°C) or Refractive index	Appearance
100	F	Cl	H	C <sub>3</sub> H <sub>7</sub> (i)	NHCH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> OCH <sub>3</sub>	101-105	White crystal
101	F	Cl	4- or 5-CH <sub>3</sub> mix.	C <sub>3</sub> H <sub>7</sub> (i)	NHCH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> OCH <sub>3</sub>	127-129	White crystal
102	F	Cl	H	C <sub>3</sub> H <sub>7</sub> (i)	NHCH <sub>2</sub> -CH< OCH <sub>3</sub> OCH <sub>3</sub>	113-114	White crystal
103	F	Cl	4- or 5-CH <sub>3</sub> mix.	C <sub>3</sub> H <sub>7</sub> (i)	NHCH <sub>2</sub> -CH< OCH <sub>3</sub> OCH <sub>3</sub>	110-113	White crystal
104	F	Cl	H	C <sub>3</sub> H <sub>7</sub> (i)	NHCH <sub>2</sub> CH< OC <sub>2</sub> H <sub>5</sub> OC <sub>2</sub> H <sub>5</sub>	118-119	White crystal
105	F	Cl	4- or 5-CH <sub>3</sub> mix.	C <sub>3</sub> H <sub>7</sub> (i)	NHCH <sub>2</sub> CH< OC <sub>2</sub> H <sub>5</sub> OC <sub>2</sub> H <sub>5</sub>	119-122	white crystal
106	F	Cl	H	C <sub>3</sub> H <sub>7</sub> (i)	NHCH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> SCH <sub>3</sub>	124-125	White crystal
107	F	Cl	4- or 5-CH <sub>3</sub> mix.	C <sub>3</sub> H <sub>7</sub> (i)	NHCH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> SCH <sub>3</sub>	127-132	White crystal
108	F	Cl	H	C <sub>3</sub> H <sub>7</sub> (i)	NHCH <sub>2</sub> CH <sub>2</sub> CN	154-156	White crystal
109	Cl	Cl	H	C <sub>3</sub> H <sub>7</sub> (i)	N< CH <sub>3</sub> CH <sub>3</sub>	106-108	White crystal
110	Cl	Cl	H	C <sub>3</sub> H <sub>7</sub> (i)	N< C <sub>2</sub> H <sub>5</sub> C <sub>2</sub> H <sub>5</sub>	96-97	White crystal
111	F	Cl	H	C <sub>3</sub> H <sub>7</sub> (i)	NHC <sub>3</sub> H <sub>7</sub> (i)	176-176.5	White crystal

The herbicidal composition of the present invention can be used either alone or in the form of a formulation according to the purpose of its use. To promote or secure the effect, it is mixed with adjuvants to make formulations such as dust, micro granule, granule, wettable powder, flowable suspension concentrates and emulsion by means of usual procedures. These formulations are used, at the time of practical application, in the form as they are or diluted with water to desired concentration.

Those adjuvants mentioned above include carriers (diluent), extending agents, emulsifiers, wetting agents, dispersing agents, fixing agents and disintegrators.

As liquid carriers there can be used water, aromatic hydrocarbons such as toluene and xylene, alcohols such as methanol, butanol and glycol, ketones such as acetone, amides such as dimethylformamide, sulfoxides such as dimethylsulfoxide, methylnaphthalene, cyclohexane, animal and vegetable oils, fatty acids and their esters, etc. As solid carriers are used clay, kaolin, talc, diatomaceous earth, silica, calcium carbonate, montmorillonite, bentonite, feldspar, quartz, alumina, sawdust, etc.

As emulsifiers or dispersing agents surfactants

are generally used. They include anionic, cationic, nonionic and amphoteric surfactants such as sodium salts of sulfated higher alcohol, stearyltrimethylammonium chloride, polyoxyethylenealkylphenylether and lauryl betaine. Wetting agents include sodium alkyl naphthalene sulfonate and ammonium polyoxyethylenealkylphenylether sulfate, fixing agents include polyvinyl alcohol, polyvinyl acetate and CMC, and disintegrators include sodium lignin sulfonate and sodium salt of lauryl sulfate.

Any type of said formulations can not only used alone, but also may be mixed with fungicides, insecticides, plant growth regulators, acaricides, soil modifying agents or nematocides and further can be used in combination with fertilizers or other herbicides.

The content of a compound (active ingredient) of the present invention in the formulations varies with types of formulation, methods of application and other conditions, but generally it is 0.5 to 95 weight %, preferably 2 to 50 weight %, while the content of adjuvants is 5 to 99.5 weight %, preferably 50 to 98 weight %, though sometimes the compound can be used alone.



To be more precise, a preferable range of the content is shown as under.

	Compound (weight %)	Adjuvant (weight %)
Dust	0.5 - 10	90 - 99.5
Emulsion	20 - 80	20 - 80
Wettable powder	20 - 80	20 - 80
Granule and micro granule	0.5 - 20	80 - 99.5
Flowable suspension concentrate	20 - 80	20 - 80

A quantity to use of the formulations varies with kinds of the active ingredient and places of application, but generally it is within the range of 1 to 100 g, preferably 3 to 75 g, of the compound per are.

The active ingredient of the present invention can be used by applying directly to weeds or the locus of the weeds.

Detailed explanation will be made below on examples formulations of the present invention and there the word "part" means part by weight.

#### Formulation Example 1 : Emulsion

35 parts of a mixture (1 : 1) of xylene and methylnaphthalene are added to 50 parts of Compound No. 14 to dissolve and the solution is further mixed with

15 parts of a mixture (8 : 2) of polyoxyethylenealkylphenylether and calcium alkylbenzenesulfonate to obtain an emulsion. It is diluted with water to use in a concentration of 0.01 to 1%.

Formulation Example 2 : Dust

5 parts of Compound No. 57 are mixed with 95 parts of clay and pulverized to obtain a dust. It is directly used for dusting.

Formulation Example 3 : Wettable powder

50 parts of Compound No. 58 are mixed with 10 parts of diatomaceous earth and 32 parts of kaolin and further uniformly blended with 8 parts of a mixture of sodium laurylsulfate and sodium 2,2'-dinaphtylmethanesulfonate, and finely pulverized to obtain a wettable powder. It is used in the form of a suspension by diluting to a concentration of 0.06 to 1%.

Formulation Example 4: Granule

5 parts of a fine dust of Compound No. 62 are extended for coating on 94.5 parts of grains (16 to 32 mesh) of silica to obtain a granule, by using a methanol solution of 0.5 parts of polyvinyl polyacetate as a binding agent in a proper mixer. The granule is a scattered directly in up-land field and paddy field.

Formulation Example 5 : Flowable suspension concentrates

40 parts of a fine powder of Compound No.111, 10 parts of ethyleneglycolmonobutylether, 10 parts of a surfactant (mixture of trioxyalkylether, polyoxyethylene-nonylphenylether and sodium alkylarylsulfonate), 3 parts of colloidal aluminium silicate hydrate and 22 parts of water are uniformly mixed and further blended under stirring in a homomixer for 20 minutes to obtain a flowable suspension concentrate. It is diluted with water for use in a concentration of 0.02 to 1%.

The excellent herbicidal activity of a compound of the present invention will be illustrated in the following test examples.

Each test was carried out on 2-replication system and the test results are given in the average value.

Test Example 1 : Pre-emergence treatment in flooded condition

A fixed amount of paddy field soil was filled in each Wagner pot sized 1/5,000 are to provide a condition similar to a paddy field and there was sown a fixed amount of seeds of barnyard grass, monochoria, toothcup, false pimpernel, water wort and umbrella plant.

In addition tubers of arrowhead were buried 1

cm under the surface of soil at the rate of 3 pieces per pot and the pot was flooded with water 3 cm deep. Then the pot was applied with a diluted solution of the compound of the present invention at a rate of 6.25 to 50 g of the compound of the present invention per are.

After three days 3 pieces of rice seedlings (variety : Nihonbare) in 2.5-leaf stage were transplanted from a nursery to each pot. Thirty days after the treatment the herbicidal activity and the phytotoxicity against paddy rice were observed. The test results were classified on the following basis as shown in Table 3.

Herbicidal activity index:

5	complete weeding
4	up to 80% weeding
3	up to 60% weeding
2	up to 40% weeding
1	up to 20% weeding
0	no effect

Phytotoxicity index:

-	no damage
+	slight damage
++	some damage
+++	moderate damage
++++	heavy damage
x	complete death

Table 2

Test Example 1: Pre-emergence treatment under flooded condition

Compound No.	Dosage g/a	Herbicidal activity				Phytotoxicity against paddy rice
		Barnyard grass	Broad leaf (1)	Umbrella sedge (2)	Arrow head	
2	25	4	5	5	5	-
	12.5	3	5	5	2	-
	6.25	2	5	5	2	-
4	25	5	5	5	3	±
	12.5	5	5	5	3	-
	6.25	4	5	5	3	-
5	25	5	5	5	5	+
	12.5	5	5	5	4.5	±
	6.25	5	5	5	4	-
8	25	5	5	5	4	±
	12.5	5	5	5	2	-
	6.25	4	5	5	2	-
9	25	5	5	5	5	+
	12.5	5	5	5	5	±
	6.25	5	5	5	5	-
10	25	5	5	5	5	++
	12.5	5	5	5	4.5	±
	6.25	5	5	5	2	±
12	25	5	5	5	3	+
	12.5	5	5	5	2	±
	6.25	5	5	5	2	-
15	25	5	5	5	4	±
	12.5	5	5	5	3	-
	6.25	5	5	5	2	-
23	25	5	5	5	4	-
	12.5	5	5	5	3	-
	6.25	5	5	5	2	-
24	25	5	5	5	5	-
	12.5	5	5	5	5	-
	6.25	4.5	5	5	5	-

Compound No.	Dosage g/a	Herbicidal activity				Phytotoxicity against paddy rice
		Barnyard grass	Broad leaf (1)	Umbrella sedge (2)	Arrow head	
27	25	5	5	5	5	+
	12.5	2	5	5	5	-
	6.25	2	5	5	5	-
29	25	5	5	5	5	-
	12.5	4.8	5	5	2	-
	6.25	4	5	5	1	-
31	25	5	5	5	5	-
	12.5	5	5	5	5	-
	6.25	4	5	5	3	-
35	25	5	5	5	5	+
	12.5	5	5	5	4	+
	6.25	4.5	5	5	3	-
37	25	5	5	5	3	-
	12.5	5	5	5	2	-
	6.25	4.5	5	5	2	-
38	25	5	5	5	5	++
	12.5	5	5	5	5	+
	6.25	5	5	5	2	±
41	25	5	5	5	5	+
	12.5	4.8	5	5	5	±
	6.25	3	5	5	4	-
46	25	5	5	5	5	+
	12.5	5	5	5	3.5	-
	6.25	5	5	5	3	-
48	25	5	5	5	3	+
	12.5	4	5	5	2	+
	6.25	3	5	5	2	-
51	25	5	5	5	5	+
	12.5	5	5	5	5	-
	6.25	4.5	5	5	4	-
52	25	5	5	5	5	±
	12.5	5	5	5	5	-
	6.25	5	5	5	4.5	-
53	25	5	5	5	5	±
	12.5	5	5	5	5	-
	6.25	5	5	5	4	-

Compound No.	Dosage g/a	Herbicidal activity				Phytotoxicity against paddy rice
		Barnyard grass	Broad leaf (1)	Umbrella sedge (2)	Arrow head	
54	25	5	5	5	5	+
	12.5	5	5	5	5	-
	6.25	5	5	5	4	-
55	25	5	5	5	4	-
	12.5	5	5	5	3	-
	6.25	5	5	5	2	-
58	25	5	5	5	5	++
	12.5	5	5	5	5	+
	6.25	5	5	5	4.5	+
62	25	5	5	5	5	++
	12.5	5	5	5	4.5	±
	6.25	5	5	5	3	-
65	25	5	5	5	5	++
	12.5	5	5	5	5	±
	6.25	5	5	5	4.5	±
67	25	5	5	5	5	++
	12.5	5	5	5	5	±
	6.25	5	5	5	4.5	±
68	25	5	5	5	2	-
	12.5	5	5	5	2	-
	6.25	5	5	5	2	-
69	25	5	5	5	5	++
	12.5	5	5	5	5	+
	6.25	5	5	5	5	±
70	25	5	5	5	5	++
	12.5	5	5	5	5	+
	6.25	5	5	5	5	+
71	25	5	5	5	5	++
	12.5	5	5	5	5	-
	6.25	5	5	5	5	-
72	25	5	5	5	5	+
	12.5	5	5	5	5	-
	6.25	5	5	5	2	-
73	25	5	5	5	5	++
	12.5	5	5	5	5	+
	6.25	5	5	5	5	+

Compound No.	Dosage g/a	Harbicial activity				Phytotoxicity against paddy rice
		Barnyard grass	Broad leaf (1)	Umbrella sedge (2)	Arrow head	
75	25	5	5	5	5	+
	12.5	5	5	5	5	-
	6.25	5	5	5	5	-
76	25	5	5	5	5	±
	12.5	5	5	5	5	-
	6.25	5	5	5	5	-
77	25	5	5	5	5	+
	12.5	5	5	5	5	±
	6.25	5	5	5	5	-
78	25	5	5	5	5	+
	12.5	5	5	5	5	±
	6.25	5	5	5	4.5	-
79	25	5	5	5	5	+
	12.5	5	5	5	5	-
	6.25	5	5	5	5	-
80	25	5	5	5	5	+
	12.5	5	5	5	5	±
	6.25	5	5	5	4.5	-
81	25	5	5	5	5	+
	12.5	5	5	5	5	+
	6.25	5	5	5	4	-
82	25	5	5	5	5	++
	12.5	5	5	5	5	+
	6.25	5	5	5	5	-
83	25	5	5	5	5	++
	12.5	5	5	5	5	+
	6.25	5	5	5	4	-
84	25	5	5	5	5	++
	12.5	5	5	5	5	+
	6.25	5	5	5	4	±
85	25	5	5	5	5	+
	12.5	5	5	5	3	-
	6.25	4	5	5	2	-
88	25	5	5	5	5	+
	12.5	5	5	5	4	-
	6.25	5	5	5	3	-

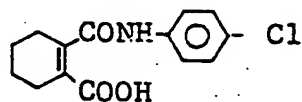


Compound No.	Dosage g/a	Harbicidal activity				Phytotoxicity against paddy rice
		Barnyard grass	Broad leaf (1)	Umbrella sedge (2)	Arrow head	
89	25	5	5	5	5	+
	12.5	5	5	5	3	±
	6.25	5	5	5	3	-
90	25	5	5	5	5	+
	12.5	5	5	5	5	-
91	25	5	5	5	5	+
	12.5	5	5	5	5	-
92	25	5	5	5	4	+
	12.5	5	5	5	3.5	-
111	25	5	5	5	5	+
	12.5	5	5	5	5	±
	6.25	5	5	5	5	-
Known compound No. 1	25	5	5	4	1	-
	12.5	4	4	3	0	-
	6.25	2	3	1	0	-
Control	25	3	4	3	0	+
	12.5	1	2	0	0	-

Remarks : (1) Broad leaf : Mixture of barnyard grass, toothcup, false pimpernel, water wort

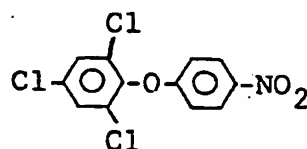
(2) Umbrella sedge : Umbrella plant

Known compound No. 1



(Japanese Patent Kokai No.44425/1973  
Example 3)

Control compound



Test Example 2 : Post-emergence treatment in flooded condition

A fixed amount of paddy field soil was filled in each Wagner pot sized 1/5,000 are to provide a condition similar to a paddy field and there was sown a fixed amount of seeds of barnyard grass, monochoria, toothcup, false pimpernel, water wort and umbrella plant.

In addition tubers of arrowhead were buried 1 cm under the surface of soil at the rate of 3 pieces per pot, three 2.5-leaf stage rice seedlings (variety : Nihonbare) were transplanted from a nursery, the pot was flooded with water 3 cm deep and then placed in a greenhouse.

When the weeds grew to reach 2 to 3-leaf stage, a diluted solution of the compound of the present invention, was applied to the flood at a rate of 12.5 to 50 g of the compound of the present invention per are.

After 30 days from the treatment with the diluted solution, the herbicidal activity was observed and obtained the results as shown in Table 3. The classification basis of the results is the same with Test Example 1.

Table 3

Test Example 2 : Post-emergence treatment in flooded condition

Compound No.	Dosage g/a	Herbicidal activity			
		Barnyard grass	Broad leaf (1)	Umbrella sedge (2)	Arrow-head
3	25	4	5	5	5
	12.5	3	5	5	5
6	25	5	5	5	5
	12.5	5	5	5	5
9	25	4	5	5	5
	12.5	3	5	5	5
13	50	5	5	5	5
	25	5	5	5	5
19	25	5	5	5	5
	12.5	5	5	5	5
25	25	4	5	5	4
	12.5	3	5	5	3
28	50	4	5	5	3
	25	2	5	5	2
36	25	5	5	5	5
	12.5	5	5	5	5
42	25	5	5	5	5
	12.5	5	5	5	5
44	25	5	5	5	5
	12.5	5	5	5	3
48	25	5	5	5	3
	12.5	3	5	5	2
50	25	5	5	5	5
	12.5	5	5	5	5
53	25	4	5	5	5
	12.5	3	5	5	3
56	25	5	5	5	5
	12.5	5	5	5	5

Compound No.	Dosage g/a	Herbicidal activity			
		Barnyard grass	Broad leaf (1)	Umbrella sedge (2)	Arrow-head
57	25 12.5	5 5	5 5	5 5	5 4
59	25 12.5	5 5	5 5	5 5	5 5
61	25 12.5	5 5	5 5	5 5	5 5
63	25 12.5	5 5	5 5	5 5	5 5
64	25 12.5	5 4.5	5 5	5 5	5 5
65	25 12.5	5 4.5	5 5	5 5	5 5
67	25 12.5	4 2	5 5	5 5	5 5
69	25 12.5	5 3	5 5	5 5	5 5
70	25 12.5	4.5 4.5	5 5	5 5	3 3
71	25 12.5	5 5	5 5	5 5	5 5
73	25 12.5	5 5	5 5	5 5	5 5
74	25 12.5	5 5	5 5	5 5	5 5
75	25 12.5	5 3	5 5	5 5	5 5
76	25 12.5	5 5	5 5	5 5	4 3
77	25 12.5	5 5	5 5	5 5	5 2
78	25 12.5	5 5	5 5	5 5	5 5
79	25 12.5	5 5	5 5	5 5	5 5
80	25 12.5	5 5	5 5	5 5	4 5

Compound No.	Dosage g/a	Herbicidal activity			
		Barnyard grass	Broad leaf (1)	Umbrella sedge (2)	Arrow-head
81	25 12.5	5 5	5 5	5 5	5 5
82	50 25	5 5	5 5	5 5	5 5
83	25 12.5	5 3	5 5	5 5	4 3
84	50 25	4.5 3	5 5	5 5	4 3
86	50 25	4 3	5 5	5 5	4 3
88	50 25	5 3	5 5	5 5	5 4
89	25 12.5	4.5 3	5 5	5 5	4.5 3
91	25 12.5	4.5 4	5 5	5 5	3 2
93	25 12.5	5 4	5 5	5 5	4 3
111	25 12.5	5 5	5 5	5 5	5 5
Known compound No. 1	25 12.5	3 0	5 5	5 5	2 0
Control A	50 25	1 0	1 0	0 0	0 0

Remarks (1)

(2) }the same as in Test Example 1

Known compound No. 1

}the same as in Test Example 1

Control A

Test Example 3: Test on perennial weeds in a paddy field

Wagner pots sized 1/5,000 are were filled with a fixed amount of paddy field soil to provide a condition similar to a paddy field and there was sown a fixed amount of seeds of bulrush. In addition tubers of mizugayatsuri and water chestnut were buried 3 cm under the surface of soil at the rate of 3 pieces per pot and then the pot was flooded with water 3 cm deep.

The pre-emergence treatment was conducted on the second day after seeds and tubers of the weeds were put into soil, while the post-emergence treatment was effected at 2-leaf stage of bulrush, 2 to 3-leaf stage of mizugayatsuri and the time when water chestnut grew 5 to 6 cm high, at each time a diluted solution of the compound of the present invention was applied to the flood at a rate of 12.5 to 50 g of the compound of the present invention per are.

The herbicidal activity was observed on 30th day after each treatment and the test results are shown in Table 4. The judging standard of the results is the same as Test Example 1.

Table 4

Test Example 3: Test on perennial weed in paddy field

Compound No.	Dosage g/a	Pre-emergence treatment			Post-emergence treatment		
		Bulrush	Mizugayatsuri	Water chesnut	Bulrush	Mizugayatsuri	Water chesnut
3	50	5	5	5	5	5	5
	25	5	5	5	5	5	4
	12.5	5	5	5	5	4	3
9	50	5	5	5	5	5	4
	25	5	5	5	3	4	3
	12.5	5	5	5	3	3	2
19	50	5	5	5	5	5	5
	25	5	5	4.5	5	5	4.5
	12.5	5	5	4	5	5	4
31	50	5	5	5	5	4	4
	25	5	5	4	5	3	3
	12.5	5	5	3	4	3	2
45	50	5	5	5	5	5	5
	25	5	5	5	5	5	5
	12.5	5	5	5	5	5	3
50	50	5	5	5	5	5	5
	25	5	5	5	5	5	4
	12.5	5	5	5	5	4	3
61	50	5	5	5	5	5	5
	25	5	5	5	5	5	5
	12.5	5	5	5	5	5	4.5
65	50	5	5	5	5	5	5
	25	5	5	5	5	5	5
	12.5	5	5	5	5	4.5	4.5
71	50	5	5	5	5	5	5
	25	5	5	5	5	5	5
	12.5	5	5	5	5	5	5
111	50	5	5	5	5	5	5
	25	5	5	5	5	5	5
	12.5	5	5	5	5	5	4

As seen in the results of Test example 1, 2 and 3, the compounds of the present invention showed remarkable herbicidal effect against the annual and perennial weeds in paddy fields in pre- and post emergence treatment.

Furthermore, it was found that the compound of the present invention showed only little phytotoxicity in pre- and post transplantation treatment.

Then the Test examples in field are shown as follows.

Test Example 4 : Pre-emergence soil surface treatment

A fixed amount of field soil was filled in a round plastic case 8 cm across and 8 cm deep, and a fixed amount of seeds of crabgrass, foxtail, pigweed, lamb's-quarters was sown followed by covering them with soil 0.5 to 1 cm thick. Then immediately a diluted solution of the compound of the present invention was applied to treat the whole surface of soil in case at a rate of 12.5 to 25 g of the compound of the present invention per are.

After the treatment the cultivation was done in a greenhouse and the herbicidal activity was observed on the 20th day. The test was carried out on 2-replication system and each average value was obtained. The judging standard of the results is the same as Test Example 1. The test results are shown in Table 5.



-40-

Table 5

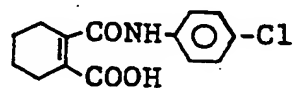
Test Example 4: Pre-emergence soil surface treatment

Compound No.	Dosage g/a	Herbicidal activity			
		Foxtail	Crabgrass	Pigweed	Buckweat
14	25	4.5	5	5	5
	12.5	3.5	5	5	5
15	25	5	5	5	5
	12.5	4	5	5	5
18	25	4.5	5	5	5
	12.5	3	5	5	5
21	25	3.5	4.5	5	5
	12.5	3	4	5	5
22	25	3	4	5	5
	12.5	2	3	5	5
24	25	4	5	5	5
	12.5	3	4	5	5
32	25	3	5	5	5
	12.5	2	4.5	5	5
42	25	3.5	5	5	5
	12.5	3	5	5	5
43	25	5	5	5	5
	12.5	5	5	5	5
44	25	5	5	5	5
	12.5	5	5	5	5
46	25	5	5	5	5
	12.5	5	5	5	5
47	25	5	5	5	5
	12.5	5	5	5	5
52	25	3	5	5	5
	12.5	2	3	5	5

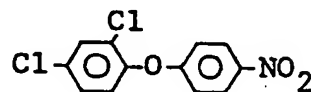
Compound No.	Dosage g/a	Herbicidal activity			
		Foxtail	Crabgrass	Pigweed	Buckweat
57	25 12.5	5 3	5 5	5 5	5 5
58	25 12.5	5 5	5 5	5 5	5 5
59	25 12.5	5 5	5 5	5 5	5 5
61	25 12.5	4.5 4	5 4	5 5	5 5
63	25 12.5	5 5	5 5	5 5	5 5
64	25 12.5	5 5	5 5	5 5	5 5
65	25 12.5	5 5	5 5	5 5	5 5
67	25 12.5	5 4.5	5 5	5 5	5 5
69	25 12.5	5 4.5	5 5	5 5	5 5
70	25 12.5	5 4.5	5 5	5 5	5 5
73	25 12.5	4 3	4.5 3	5 5	5 5
74	25 12.5	4 3	4 3	5 5	5 5
75	25 12.5	4.5 4	5 5	5 5	5 5
76	25 12.5	5 5	5 5	5 5	5 5
77	25 12.5	5 5	5 5	5 5	5 5
78	25 12.5	5 5	5 5	5 5	5 5
79	25 12.5	5 5	5 5	5 5	5 5
80	25 12.5	5 5	5 5	5 5	5 5

Compound No.	Dosage g/a	Herbicidal activity			
		Foxtail	Crabgrass	Pigweed	Buckweat
81	25	5	5	5	5
	12.5	5	5	5	5
82	25	5	5	5	5
	12.5	5	5	5	5
83	25	5	5	5	5
	12.5	5	5	5	5
86	25	4.5	4.5	5	5
	12.5	4.5	4	5	5
88	25	5	5	5	5
	12.5	5	5	5	5
91	25	5	5	5	5
	12.5	5	5	5	5
111	25	5	5	5	5
	12.5	5	5	5	5
Known compound No. 1	25	2	2	5	4
	12.5	1	0	5	2
Control B	25	4	4.5	5	5
	12.5	2	3	5	5

Known compound No. 1



Control B



Test Example 5 : Post-emergence treatment

A fixed amount of field soil was filled in a round plastic case 8 cm across and 8 cm deep, and a fixed amount of seeds of foxtail, pigweed was sown. When they grew up to 3 to 4-leaf stage, a wettable powder containing the compound of the present invention was sprayed on the body of plants after diluting it at a rate of 12.5, 25 or 50 g of active ingredient per are.

The test was conducted on 2-replication system. Twenty days after the treatment the test results were observed on the same judging standard and the results are shown in Table 6.

Table 6

Test Example 5 : Post-emergence treatment

Compound No.	Dosage g/a	Herbicidal activity		Compound No.	Dosage g/a	Herbicidal activity	
		Foxtail	Pigweed			Foxtail	Pigweed
4	50	5	5	62	50	5	5
	25	4	5		25	4.5	5
	12.5	3	5		12.5	3.5	5
12	50	5	5	63	50	5	5
	25	5	5		25	4	5
	12.5	4	5		12.5	3	5
21	50	5	5	66	50	4	5
	25	3.5	5		25	3	5
	12.5	3	5		12.5	2	5
34	50	4.5	5	67	50	3.5	5
	25	4	5		25	3	5
	12.5	2	4.5		12.5	2	4.5
41	50	5	5	69	50	5	5
	25	4.5	5		25	4	5
	12.5	3.5	3.5		12.5	3	5
43	50	5	5	70	50	5	5
	25	3	5		25	4	5
	12.5	2	5		12.5	3	5
56	50	5	5	71	50	5	5
	25	5	5		25	5	5
	12.5	4	5		12.5	4	5
57	50	5	5	72	50	4	5
	25	4	5		25	3	5
	12.5	3	5		12.5	2	5
58	50	5	5	74	50	4	5
	25	5	5		25	2.5	5
	12.5	4.5	5		12.5	2	5
59	50	5	5	75	50	4	5
	25	5	5		25	3	5
	12.5	5	5		12.5	2	5
76	50	5	5	93	50	3.5	5
	25	5	5		25	3	5
	12.5	4.5	5		12.5	2	5

Compound No.	Dosage g/a	Herbicidal activity		Compound No.	Dosage g/a	Herbicidal activity	
		Foxtail	Pigweed			Foxtail	Pigweed
77	50	5	5	111	50	5	5
	25	5	5		25	5	5
	12.5	5	5		12.5	5	5
78	50	5	5	Known compound No. 1	50	0	5
	25	5	5		25	0	4
	12.5	5	5		12.5	0	3
79	50	5	5	Control B	50	5	5
	25	5	5		25	3	5
	12.5	5	5		12.5	2	4.5
80	50	5	5				
	25	5	5				
	12.5	5	5				
81	50	5	5				
	25	4.5	5				
	12.5	4.5	5				
82	50	5	5				
	25	3	5				
	12.5	2	5				
83	50	5	5				
	25	5	5				
	12.5	3	5				
89	50	5	5				
	25	4	5				
	12.5	3	5				
91	50	4	5				
	25	3	5				
	12.5	2	5				

Test Example 6 : Phytotoxicity against crops

A fixed amount of field soil was filled in a plastic vessel sized 23 cm x 4.5 cm x 12.5 cm and a fixed amount of seeds of soybean, cotton, corn, wheat, sunflower and rice was sown followed by 3-cm thick covering with soil.

Then immediately a diluted solution of the compound of the present invention was sprayed on the soil surface with a small sprayer at the rate of 25 to 50 g of the compound of the present invention.

After the treatment the crops were grown in a greenhouse and 20 days later the degree of phytotoxicity against each crop was observed. The test was carried out on 2-replication system and each average value was obtained.

The judging standard of test results is the same as Test Example 1 and the result are shown in Table 7.

Table 7

Test Example 6 : Phytotoxicity against crops

Compound No.	Dosage g/a	Phytotoxicity against crops					
		Soybean	Cotton	Corn	Wheat	Rice	Sunflower
15	50 25	- -	- -	- -	- -	- -	- -
18	50 25	- -	- -	- -	- -	- -	- -
24	50 25	- -	- -	- -	- -	- -	- -
43	50 25	- -	- -	- -	- -	- -	- -
46	50 25	- -	- -	- -	- -	- -	- -
47	50 25	- -	- -	- -	- -	- -	- -
58	50 25	- -	- -	- -	- -	- -	- -
59	50 25	- -	- -	- -	- -	- -	- -
63	50 25	- -	- -	- -	- -	- -	- -
64	50 25	- -	- -	- -	- -	- -	- -
70	50 25	- -	- -	- -	- -	- -	- -
73	50 25	- -	- -	- -	- -	- -	- -
75	50 25	- -	- -	- -	- -	- -	- -
76	50 25	- -	++ +	+ -	+ -	+ -	- -
78	50 25	- -	+ -	- -	- -	+ -	- -
80	50 25	+ -	+ -	- -	+ -	+ -	- -

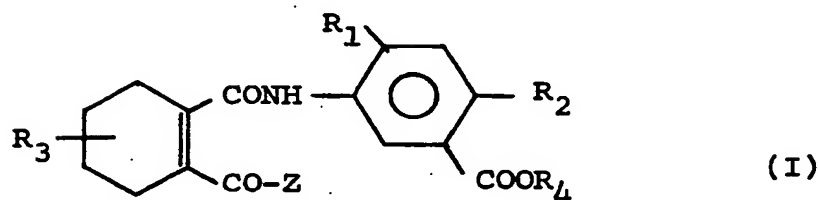


Compound No.	Dosage g/a	Phytotoxicity against crops					
		Soybean	Cotton	Corn	Wheat	Rice	Sunflower
81	50 25	+	++	+	-	+	-
82	50 25	+	++	-	-	+	-
83	50 25	+	+	-	-	+	-
84	50 25	-	+	-	-	+	-
86	50 25	+	++	-	-	+	-
88	50 25	+	+	+	-	+	-
92	50 25	+	++	+	+	-	-
93	50 25	-	+	+	-	+	-
111	50 25	-	+	-	-	-	-
Known compound No. 1	50 25	-	-	-	-	-	-
Control B	50 25	++ +	+++ ++	+++ ++	+++ ++	++ +	+ -

As obvious from the results of Test Examples 4 and 5, the compound of the present invention proves to show very good herbicidal activity both in pre-emergence and post-emergence treatments of main weeds in the field. On the other hand, it is clear from the results of Test Example 6 that the compound of the present invention has no phytotoxicity against crops and is a suitable herbicide for use in farmlands.

CLAIMS

1. An N-substituted-3,4,5,6-tetrahydrophthalamic acid derivative of the formula:

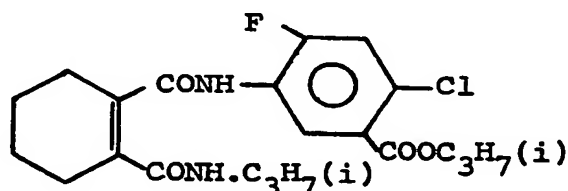


wherein  $R_1$  is hydrogen or halogen;  $R_2$  is halogen;  $R_3$  is 5 hydrogen or  $C_1$ - $C_4$  alkyl;  $R_4$  is hydrogen or  $C_1$ - $C_8$  alkyl which may be substituted by halogen or  $C_1$ - $C_4$  alkoxy; and Z is hydroxy,  $C_1$ - $C_8$ -primary or secondary alkylamino,  $C_2$ - $C_5$ -alkenyl amino,  $C_3$ - $C_7$  cycloalkylamino, furfurylamino, benzylamino, morpholino which may be substituted by methyl, 10 or  $C_2$ - $C_4$  alkylamino which may be substituted by phenyl,  $C_1$  or  $C_2$ -alkylamino,  $C_1$ - $C_4$ -alkoxy,  $C_1$ - $C_4$ -alkylthio or cyano; or an alkali metal salt thereof.

2. A compound according to claim 1 wherein  $R_1$  is chloro or fluoro,  $R_2$  is chloro or bromo,  $R_3$  is hydrogen 15 or methyl,  $R_4$  is  $C_2$  or  $C_3$ -alkyl and Z is hydroxy,  $C_1$ - $C_3$ -alkylamino or  $C_1$ - $C_3$ -alkoxy- $C_2$  or  $C_3$ -alkylamino.

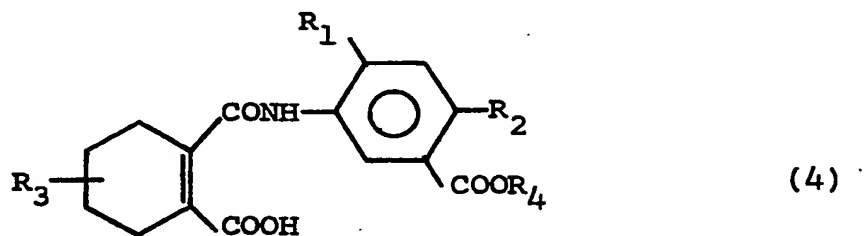
3. A compound of claim 2 wherein  $R_1$  is fluoro,  $R_2$  is chloro or bromo,  $R_3$  is hydrogen,  $R_4$  is isopropyl and Z is  $C_1$ - $C_3$ -alkylamino or  $C_1$ - $C_3$ -alkoxy- $C_2$  or  $C_3$ - 20 alkylamino.

4. A compound according to claim 3 of the formula:

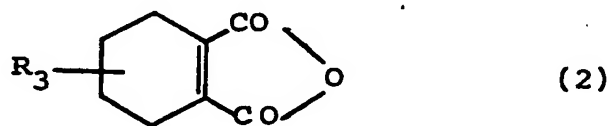


wherein  $\text{C}_3\text{H}_7(\text{i})$  represents isopropyl.

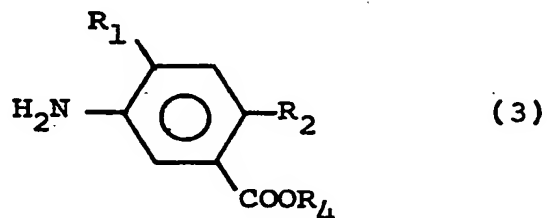
5. A process for the preparation of an N-substituted-3,4,5,6-tetrahydrophthalamic acid derivative of the formula (4):



wherein  $R_1, R_2, R_3$  and  $R_4$  are as defined in claim 1, or an alkali metal salt thereof, which process comprises reacting a 3,4,5,6-tetrahydrophthalic anhydride of the formula (2):

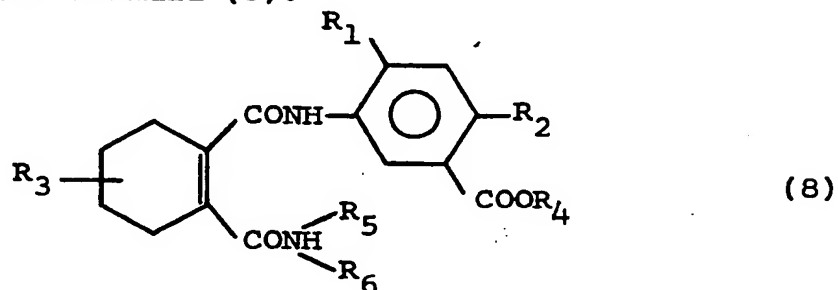


wherein  $R_3$  is as defined above, with a m-aminobenzoic acid derivative of the formula (3):

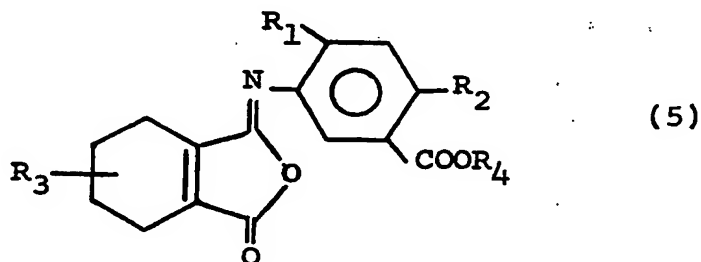


wherein  $R_1, R_2$  and  $R_4$  are as defined above, and, if desired, converting the resulting compound of formula (4) into an alkali metal salt thereof.

6. A process for the preparation of an N-substituted-3,4,5,6-tetrahydrophthalamic acid derivative of the formula (8):



5 wherein  $R_1, R_2, R_3$  and  $R_4$  are as defined in claim 1,  $R_5$  is hydrogen or  $C_1-C_8$ -alkyl and  $R_6$  is hydrogen,  $C_1-C_8$ -alkyl,  $C_2-C_5$ -alkenyl,  $C_3-C_7$ -cycloalkyl, furfuryl, benzyl or  $C_2-C_4$ -alkyl which may be substituted by phenyl,  $C_1$  or  $C_2$ -alkylamino,  $C_1-C_4$ -alkoxy,  $C_1-C_4$ -alkylthio or cyano or  $R_5$  and  $R_6$  together with the nitrogen atom to which they are attached form a morpholino group which may be substituted by methyl, which process comprises reacting a compound of the formula (5):



15 wherein  $R_1, R_2, R_3$  and  $R_4$  are as defined above, with a compound of the formula (7):



wherein  $R_5$  and  $R_6$  are as defined above, in an aqueous or organic solvent.

Formula (8):

(8)

\*C1=CC=C(C(=O)O\*)C=C1N2C(=O)C3C=CC=C(C3)C2 (6)
$$\text{HN} \begin{cases} \text{R}_5 \\ \text{R}_6 \end{cases} \quad (7)$$

9. A herbicidal composition which comprises from 0.5 to 95 weight % of an N-substituted-3,4,5,6-tetrahydro-  
20 phthalamic acid derivative of the formula (1) or an alkali metal salt thereof as claimed in any one of claims 1 to 4 or which has been produced by a process as claimed in any

one of claims 5 to 8, and from 5 to 99.5 weight % of adjuvant(s).

10. A method of controlling the growth of weeds at a locus, which method comprises applying to the locus a herbicidally effective amount of a derivative of N-substituted-3,4,5,6-tetrahydrophthalamic acid of the formula (1) or an alkali metal salt thereof as claimed in any one of claims 1 to 4 or which has been produced by a process as claimed in any one of claims 5 to 8.